Ag(I)-ASSISTED HYDROLYSIS OF MESTRANOL METHANESULFONATE INTO EPIMESTRANOL

H. Westmijze, H. Kleijn and P. Vermeer*

(Department of Organic Chemistry of the State University, Croesestraat 79, 3522 AD UTRECHT, The Netherlands)

and L.A. van Dijck

(Analytical R & D Labs, Organon Scientific Development Group, 5340 BH OSS, The Netherlands)

Summary: Epimestranol (1b) is obtained in 80% yield from mestranol (1a) by converting 1a into its methanesulfonate 2 and treating 2 with silver(I) nitrate in a mixture of tetrahydrofuran and water.

Although the synthesis of 3-methoxy-17 α -ethynyl-17 β -hydroxyestra-1,3,5(10)-triene (1a, mestranol) by ethynylation of the corresponding 17-ketone is well known, 1 an efficient method for the preparation of its 17-epimer (1b, epimestranol) is not available in the literature.

In three papers a low yield synthesis for 1b has been reported, viz. epimerization of mestranol acetate on alumina (yield of 1b: 5%), 2^a , 2^b and conversion of 1a via enyme 3 and epoxide 4 into 1b (eq. 1; overall yield of 1b: 6%). 2^b , 3

MeO (1a) (1b)
$$C \equiv CH$$
 $C \equiv CH$
 $C \equiv CH$

In view of our interest in $10^{4.5}$ we decided to search for a more attractive route to 10^{10} . In this paper we wish to present a simple, high yield synthesis of 10^{10} from readily available 10^{10} . The strategy we followed to convert 10^{10} into 10^{10} is outlined in eq. 2. As is shown in eq. 2, the method involves the preparation of mestranol methanesulfonate 2. This compound was readily obtained by subsequent treatment of 10^{10} with n-butyllithium and methanesulfonyl chloride. Although sulfonate esters derived from tertiary alcohols are known generally to be unstable, 2^{10} could be isolated as a crystalline compound [yield: >98%, 10^{10} NMR (CDC1 $_3$, 10^{10} Me $_4$ Si = 0 ppm) data found for 10^{10} could 10^{10} data 10^{10} could 10^{10} data 10^{10} data 1

Initial attempts to convert 2 into 1b by H⁺-catalysed hydrolysis in, for instance, water containing tetrahydrofuran (THF) were disappointing because of formation of large amounts of enyne 3, a compound which is formed as the main product during the epimerization of mestranol acetate on alumina $(cf.^2)$. In contrast, the Ag⁺-catalysed hydrolysis of 1a in THF-H₂O furnished the desired epimer 1b in excellent yield, together with a small amount of enyne 3 (Ratio 1b: 3 \sim 95:5).

Compound 1b could easily be purified from 3 by treating the crude product with boiling hexane [yield of pure 1b: 80%; $\left[\alpha\right]_{D}^{20}$ (in CH_2Cl_2) + 71.3°; m.p. 139.5-140.0°C (reported 135-136°C); ^1H NMR (CDCl $_3$, $^3\text{Me}_4\text{Si}$ = 0 ppm): $^3\text{Me}_3\text{O}$ (13-Me), 2.47 (HC=), 3.78 (MeO), 6.5-7.3 (aromatic protons)]

The experimental procedure is as follows: To a stirred solution of 1a (0.020 mol) and LiBr (0.020 mol) in THF (70 ml) is added at -60°C n-butyllithium (0.020 mol; 1.5 M-solution in n-hexane) followed, after 30 min, by methanesulfonyl chloride (0.020 mol). After stirring for 30 min at -60°C, the reaction mixture is poured into an aqueous NH₄Cl solution (200 ml). Product 2 is isolated by extraction with CH_2Cl_2 (2 x 100 ml), drying of the combined extracts with MgSO₄, and, after the addition of pyridine (1 ml), evaporation of the solvent in vacuo at 0°C. The crude crystalline compound is washed with dry diethyl ether (1 x 30 ml). Subsequently, sulfonate 2 is stirred with AgNO₃ (0.3 g) in a mixture of THF (30 ml) and distilled H₂O (5 ml) for 2 hrs. at 20°C. The resulting reaction mixture is then poured into an aqueous NH₄Cl solution (200 ml), containing NaCN (2g); crude 1b is isolated as described for 2. The small amount of 3 is removed by treating crude 1b with boiling n-hexane (100 ml), cooling the resulting mixture to 20°C, and filtering off crystalline 1b. The presented method will probably also be useful for the conversion of other 17α -ethynyl- 17β -hydroxysteroids into their interesting 17-epimers.

References

- 1. F.B. Colton, U.S. Patent 3,666,769 (1954).
- 2ª.R.M. Kanojia, K, Yarmchuck, and I. Scheer, J. Org. Chem. 39 (1974) 2304.
- 2^b.R.M. Kanojia, G.O. Allen, J.M. Killinger, and J.L. McQuire, J. Med. Chem. 22 (1979) 1538.
- 3. L.A. van Dijck, B.J. Lankwerden, and J.G.C.M. Vermeer, Recl. Trav. Chim. Pays-Bas 96
- 4. L.A. van Dijck, B.J. Lankwerden, and J.G.C.M. Vermeer, ibid 98 (1979) 553.
- 5. H. Westmijze and P. Vermeer, Tetrahedron Lett. (1979) 4101.

(Received in USA 15 april 1980)